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bond of an alkane?

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$\text{CH}_3\text{C}\equiv\text{N}$  and the bond

angle  $\text{CCN}$  are a.  $\text{sp}^2$ ,

$180^\circ$ . b.  $\text{sp}$ ,  $180^\circ$ . c.  $\text{sp}^2$ ,

$120^\circ$ . d.  $\text{sp}^3$ ,  $109^\circ$ . 2.

Which of the following

statements about an  $\text{sp}$



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hybridized carbon is FALSE? a. It is divalent. b. It forms bonds that are linear. c. It has two p orbitals. d.

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constitutional isomers

B. stereoisomers C.

identical D. not isomers;  
different compounds

entirely Answer: A

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