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Chapter 13 – Gases 195 Exercise 13.3 – Equation Stoichiometry: Iron is combined with carbon in a series of reactions to form pig iron, which is about 4.3% carbon.  $2C + O_2 \rightarrow 2CO$   $Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$   $2CO + C \rightarrow 3C$  (in iron)  $CO_2$  Pig iron is easier to shape than pure iron, and the presence of carbon lowers its melting point

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Name Date Class 76 Chemistry: Matter and Change • Chapter 13 Study Guide for Content Mastery Section 13.3 Liquids and Solids In your textbook, read about liquids and solids. In the space at the left, write true if the statement is true; if the statement is false, change the italicized word or phrase to make it true.

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Chemistry Chapter 13 study guide questions question what is pressure? answer a property of all gas. It exerts pressure on (pushes against) the sides question what are the instruments that are used to

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CHAPTER 13 Table Of Contents Click a hyperlink to view the corresponding slides. •State the relationships among pressure, temperature, and volume of a constant amount of gas.

Chemistry: Matter and Change

196 Study Guide for An Introduction to Chemistry Chapter 13 Map Chapter Checklist Read the Review Skills section. If there is any skill mentioned that you have not yet mastered, review the material on that topic before reading this chapter. Read the chapter quickly before the lecture that describes it.

Chapter 13 - Gases - An Introduction to Chemistry

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