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~~and atoms Basic~~

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Constituents of

Matter

Basic Chemistry

for Biology,

Part 1: Atoms

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Elements, Atoms,

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number 2 titled

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based tool,
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and paste it
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box. Terms in
this set (40)

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Answers These 4
elements form
approximately
96% of almost
all life forms.

C, H, O, N The
atomic number of
an atom

represents The
number of

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The atomic mass of an atom

represents The number of protons +the number of neutrons Oxygen has an

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Answers*

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Chemistry. The

Human Body in

Health and

Illness, 4th

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PLAY. matter.

anything that

occupies space

and has weight.

Exists as solid,

liquid and gas.

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Undergoes
physical and
chemical change.
physical change.

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BIOCHEMISTRY:

THE COMPOSITION
OF LIVING MATTER

14. Use the key
choices to
identify the
substances
described in the
following
statements.

Insert the

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appropriate
letter(s) or
corresponding
term(s) in the
answer blanks.

B. Base(s) C.

Buffer. D.

Salt(s) 3-

Substances that
ionize in water;
good
electrolytes 4.

answer key wb

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chpt 2 14-17 |

Hydroxide |

Chemistry

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Answer Key. .

challenging the

brain to think

better and

faster can be

undergone by

some ways.

Experiencing,

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listening to the
supplementary
experience,
adventuring,
studying,
training, and
more practical
activities may
encourage you to
improve.

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Answer Key

REVIEW OF BASIC
CHEMISTRY ANSWER

KEY 1. Name the
following
elements.

Spelling counts:

H hydrogen S

sulphur Cl

chlorine C

carbon N

nitrogen Na

sodium Pb lead

Ag silver 3.

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Using a periodic
table, record
the atomic
number for the
following
elements:

R BASIC

CHEMISTRY ANSWER

KEY - Just Only

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Julia Publisher
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6 7 Unit 2:

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ANSWER KEY

MATTER AND

ENERGY Chapter 2

. Select all phrases that apply to each of the following statements and insert the letters in the answer blanks.

1. The energy

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located in the
bonds of food
molecules: A. is
called thermal
energy B. is a
form of
potential energy
AB, cD -L- 2.
Heat is:

*Mr. Burghardt
Ponderosa High
School Shingle
Springs, CA
Page 33/46*

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Chapter 2 – The
Structure of
Matter and the
Chemical

Elements 15 27.

When an atom
gains one or
more electrons,
it then has more
electrons than
protons and more
minus charge
than plus
charge. Thus it

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becomes an
anion, which is
an ion with a
negative charge.

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Structure of
Matter and the
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Elements*

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the basic unit
of matter:

nucleus: the
center of the

atom: electron:

a negatively
charged

particle:

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Answer Key
element: a pure
substance that
consists

entirely of one
type of atom:

isotope: atoms
of the same
element that
differ in the
number of

neutrons they
contain:

compound: a
substance formed

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Answer Key

by the chemical
combination of
two or more
elements in
definite
proportions

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2: The Chemistry
of Life*

Vocabulary

Review

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for Class 9

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Science Chapter

2 Is Matter

Around Us Pure

provides

detailed answers

and explanations

for the exercise

questions

provided in the

chapter. It

covers even

minute details,

which will help

you in clearing

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fundamentals to
help students
with ...

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Matter Around

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Chemistry by
Rebekah Branka |
TpT. This
chapter focuses
on the following
concepts:-
Concepts of
matter and

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Chemistry

Composition of
matter

(elements,
atoms, atomic
structure,
identifying
elements) -

Molecules and
compounds -

Chemical bonds
and chemical
reactions -

Inorganic

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Chemistry
Compounds
(salts,
Answer Key
acids/bases,
water) - Organic
compounds
(carboh...

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