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Answers to Practice Problems. 1) This is a basic electrophilic hydration. 2) The answer is additional side products, but the major product formed is still the same (the product shown). Depending on the temperatures used, the cyclopropane may open up into a straight chain, which makes it unlikely that the major product will form (after the reaction, it is unlikely that the 3º carbon will remain as such).

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K. Peter C. Vollhardt. K. Peter C. Vollhardt was born in Madrid, raised in Buenos Aires and Munich, studied at the University of Munich, got his Ph.D. with Professor Peter Garratt at the University College, London, and was a postdoctoral fellow with Professor Bob Bergman (then) at the California Institute of Technology. He moved to Berkeley in 1974 when he began his efforts toward the ...

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I like Vollhardt's orgo texts, so I bought this as a study aid during my organic chemistry sequence. I used it for most chapters, but I can't say that it made a huge difference. The layout of the book is ok, and the questions are relevant and appropriately challenging for a non-chemistry major orgo course, but I feel I could have gotten an equal grade without it.

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Organic and Organometallic Chemistry – Transition metals and novel synthetic methodology enable the synthesis of complex natural and unnatural products. Professor Vollhardt and his coworkers are interested in the application of transition metals to organic synthesis. The areas under investigation range from the assembly of novel oligometallic arrays to the use of cobalt, palladium, and nickel in the construction of complex organic molecules.

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Moving back to New York, he worked with the late Professor Nicholas J. Turro at Columbia University, studying photochemical and photophysical processes of organic compounds for his Ph.D. thesis. He first met co-author Peter Vollhardt in the 1970s when they both were doing postdoctoral work in Professor Robert Bergman's laboratory at Cal Tech.

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