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Chapter End Questions. Question
1 Which of the statements about
the reaction below are incorrect ?
$$2 \text{PbO}(s) + \text{C}(s) \rightarrow 2\text{Pb}(s) + \text{CO}_2(g)$$

(a) Lead is getting reduced.

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(b) Carbon dioxide is getting oxidised. (c) Carbon is getting oxidised. (d) Lead oxide is getting reduced. (i) (a) and (b) (ii) (a) and (c)

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How do you determine whether a chemical reaction has taken place? Answer: Following observations helps us to determine whether a chemical reaction has taken place

- Change in state
- Change in colour
- evolution of a gas
- Change in

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Foundations Exercise 2. Write the Balanced equation for the reaction of iron with steam.

Answer: $3\text{Fe(s)} + 4\text{H}_2\text{O(g)} \rightarrow \text{Fe}_3\text{O}_4\text{(s)} + 4\text{H}_2\text{(g)}$ Question 3.

What is quicklime? Write one use of quicklime. Answer: Calcium oxide is called lime or quicklime.

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1.1: Chemical Equations: 1.1.1:
Writing A Chemical Equation:
1.1.2: Balanced Chemical
Equations: 1.2: Types Of Chemical

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Reactions: 1.2.1: Combination

Reaction: 1.2.2: Decomposition

Reaction: 1.2.3: Displacement

Reaction: 1.2.4: Double

Displacement Reaction: 1.3: Have

You Observed The Effects Of

Oxidation Reactions In Everyday

Life? 1.3.1: Corrosion: 1.3.2:

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Answers. 1. oxidation, reduction
2. reduction, oxidation 3. salts 4.
exothermic reactions 5.
displacement reaction 6. BHA and

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BHT 7. Corrosion 8. SO₃ 9. BaSO₄

10. (a) $x \rightarrow (s)$; $y \rightarrow (aq)$; (b) $x \rightarrow$

2Ag. We hope the given MCQ

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you.

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Equations Class 10 Important
Questions Short Answer Type I.
Question 1. Write the balanced
chemical equation for the

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following reaction and identify the type of reaction and define it.

'Iron III oxide reacts with Aluminium and gives molten iron and aluminium oxide'. (2012)

Answer: $\text{Fe}_2\text{O}_3 (\text{s}) + 2\text{Al} (\text{s}) \rightarrow 2\text{Fe} (\text{l}) + \text{Al}_2\text{O}_3 (\text{l}) + \text{Heat}$

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- Challenge Problems - Page 40:
111 Answer a) Percentage error =
2.222 % b) Percentage error =
2.233 % c) Percentage error =

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with Answers. Choose the correct
option. Question 1.

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Teachers Chemicals Reactions &
Equations - Chapter 1 Student
Worksheet Time:20 minutes
Answer the following questions :

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Q1. Write the names of the reactants taking part in the chemical reaction during the demonstration. Q2. Write the chemical formulae of both the reactants. Q3.

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- In a chemical reaction, the atoms and molecules produced by the reaction are called products.
- In a chemical reaction, only the atoms present in the reactants can end up in the products. No new atoms are

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created, and no atoms are
destroyed. • In a chemical
reaction, reactants contact each
other, bonds between atoms in
the reactants

Chapter 6, Lesson 1: What is a Chemical Reaction?

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NCERT Questions & Answers.

Question 1. Mention the factors that affect the rate of a chemical reaction. Answer: concentration of reactants; temperature; nature of reactants and products; exposure to light (radiation) presence of catalysts; Question 2.

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A reaction is second order with respect to a reactant.

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